

CBM Reservoir

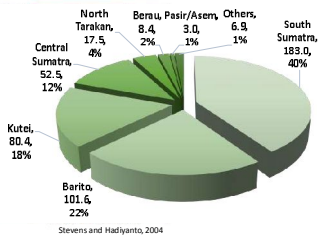
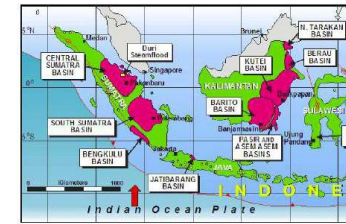
Adsorption (Gas content)

Ferian Anggara

Outline

- Overview
- Adsorption (Gas content)
- Porosity
- Permeability (Cleft system)
- Gas Flow
- Reserve Analysis
- Dynamic reservoir
- Enhanced recovery

Coal and CBM resources in Indonesia



Coal bed methane (CBM):
337-453 Trillion cubic feet

Stevens and Hafiyanto, 2004

Coal resources and reserves

Class	Quality Calorific Value (cal/gram)	Resources Million tons		Reserve (Million tons)			
		Total	%	Probable	Proven	Total	%
Low	< 5,100	21,227.63	20.18%	7,603.88	1,105.40	9,709.28	41.21%
Medium	5,100 - 6,100	69,726.02	66.29%	7,063.52	2,904.41	9,967.93	47.17%
High	6,100 - 7,100	13,220.61	12.57%	861.73	1,410.44	2,272.17	10.75%
Very High	> 7,100	1,013.39	0.96%	73.29	109.38	182.47	0.87%
Total		105,187.44	100.00%	15,602.41	5,529.43	21,131.85	100.00%

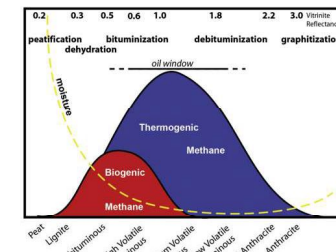
Geological agency, 2010

Taylor et al., 1998:
LRC: Low rank coal → < 7000 cal/gram
HRC: High rank coal → > 7000 cal/gram

What is Coal Gas?

- This gas was considered one of the many **hazards** of coal mining
- **Recently:** unconventional resources
- Gas is generated during maturation of organic matter into coal and by microbes residing in a coal
- Coal deposits of all geologic ages have generated gas

Schematic showing biogenic and thermogenic gas generation



Moore, 2012

Typical of CBM Production Curve

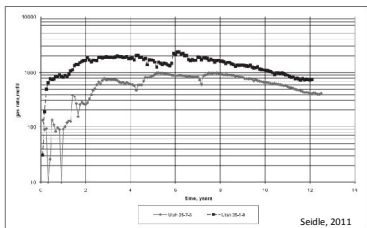
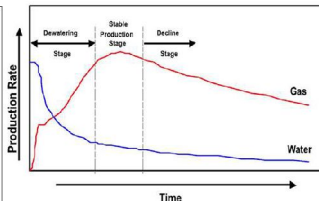


Fig 1-1. Examples of the negative decline—Bita Basin, Form coal



Conventional Gas Vs. CBM

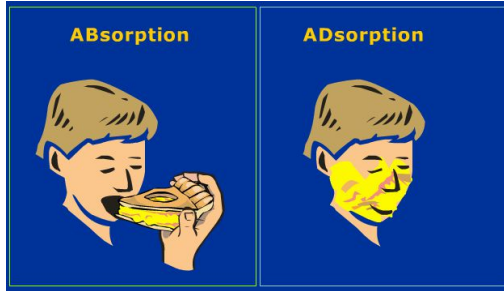
Conventional Gas	Coalbed
Darcy flow of gas to wellbore.	Diffusion through micropores by Fick's Law. Darcy flow through fractures.
Gas storage in macropores: real gas law.	Gas storage by adsorption on micropore surfaces.
Production: schedules according to set decline curves.	Initial negative decline
Gas content from logs.	Gas content from cores. Cannot get gas content from logs.
Gas to water ratio decreases with time	Gas to water ratio increases with time in latter stages.
Inorganic: reservoir rock.	Organic: reservoir rock.
Hydraulic fracturing may be needed to enhance flow.	Hydraulic fracturing required in most of the basins except the eastern part of the Powder River basin where the permeability is very high. Permeability dependent on fractures.
Macropore size: ³ 1 μ to 1 mm	Micropore size: ³ <5Å ³ to 50Å ³
Reservoir and source rock independent.	Reservoir and source rock same.
Permeability not stress dependent.	Permeability highly stress dependent.
Well interference detrimental to production.	Well interference helps production. Must drill multiple wells to develop.

Levine, 1990

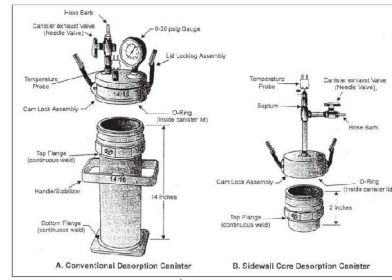
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ADsorption vs ABsorption



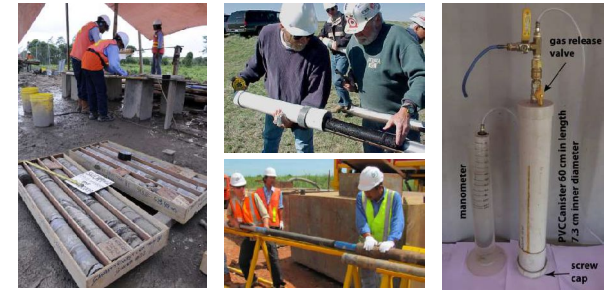
Management Reservoir



Seidle, 2011

Management Reservoir

Sampling



Moore, 2012
Chiper coal, 2015

Management Reservoir

Gas desorption measurements

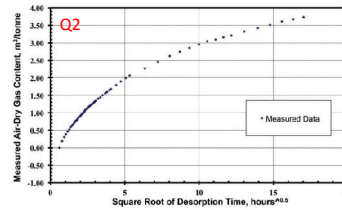
- 1. Lost gas (Q1)**
is the calculated amount of gas that has potentially been lost from the time the core leaves the reservoir at depth to when it is placed within a canister at the surface
- 2. Measured gas (Q2)**
the gas which is directly desorbed from the canister and measured using a manometer
- 3. Residual gas (Q3)**
 - After a sample is either 'fully' desorbed or when the operator decides that enough has been measured to achieve the desired goals of the sampling, the coal is removed from the canister
 - Crushing coal to approximately 250 μm size particles, which has gas tight seals and a gas lead from which the gas evolved can be removed for measurement.

$$\text{Total gas volume} = Q1 + Q2 + Q3$$

Moore, 2012

Management Reservoir

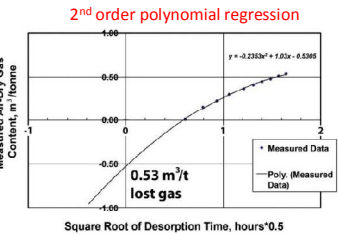
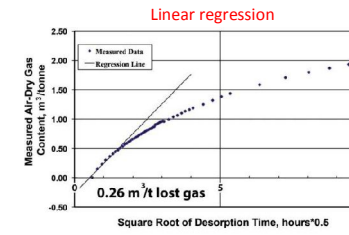
Q2 & Q3 calculation



Moore, 2012

Management Reservoir

Q1 calculation



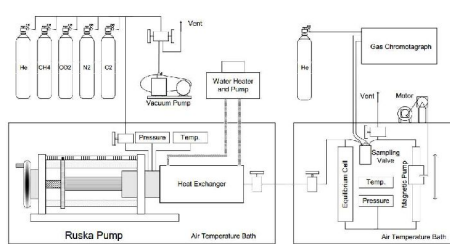
Moore, 2012

Management Reservoir

Adsorption test



PVT Apparatus, Kyushu University-Japan



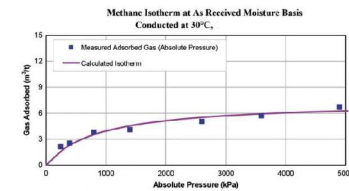
Gasem et al., 2009

Management Reservoir

Adsorption test

Client	Micah Global CBM	Gas	Methane
Sample	Brandon Basin	Sample No	Rollie-20-7A
Location	Texas Mine	Formation	Jasper Fm
		Depth (m)	410
		Testing	N/A
		Nothing	N/A
		Density (gm³)	1.33

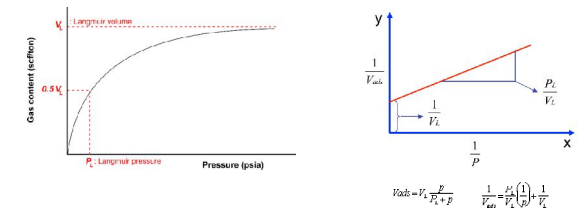
Adsorption Equation		Proximate Analysis (%)	
$V = \frac{a(P - P_0)}{b + P - P_0}$		Inherent Moisture	3.3
where:		ASH	1.1
V = Gas content (m ³ /t)		Volatile	32.8
P = Pressure (kPa)			
a = Langmuir Volume (m ³ /t)			
b = Langmuir Pressure (kPa)			



Test Summary		Gas Adsorbed (m³/t)	
Absolute Pressure (kPa)	Measured	Calculated	
101	0.00	0.77	
221	2.09	1.65	
401	2.50	2.32	
601	3.74	3.53	
1401	4.07	4.54	
2601	5.01	5.51	
3401	5.62	5.92	
4901	6.62	6.25	

Management Reservoir

Langmuir parameters



$$V_{\text{ads}} = V_L \frac{P}{P_L + P} \quad \frac{1}{V_{\text{ads}}} = \frac{1}{V_L} \left(\frac{P_L}{P} + 1 \right) = \frac{1}{V_L} + \frac{P_L}{V_L P}$$

Management Reservoir

Gas saturation



- Desorption Isotherms**
 - ❖ "slow", "fast" or "in between"
 - ❖ reservoir temperature
- Adsorption Isotherms**
 - ❖ reservoir temperature
 - ❖ fresh samples!

$$\%g = 1 - ((a-d)/a) * 100$$

where,
 %g, is gas saturation
 a, is maximum gas holding capacity [adsorption]
 d, is total measured gas [desorption]

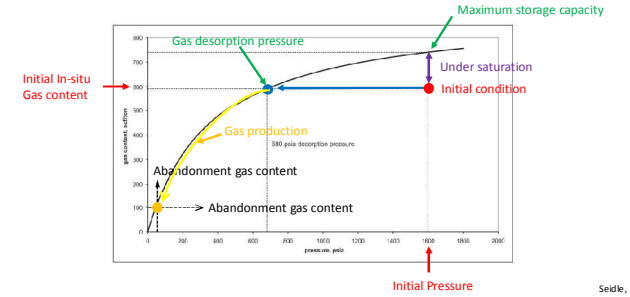
Moore, 2010



• But different types of coal can hold more or less gas (i.e. different coals have different holding capacities)

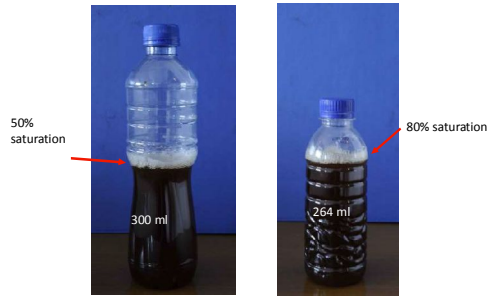
Moore, 2010

Gas saturation and desorption pressure



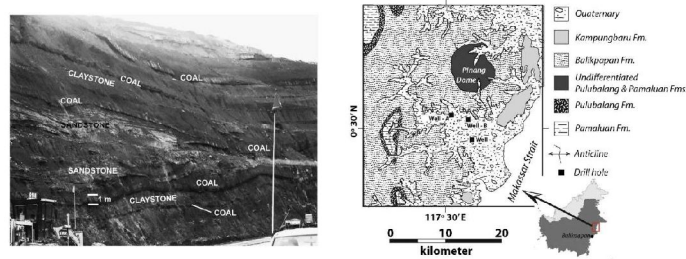
Seidle, 2011

Which deposit would you pick?



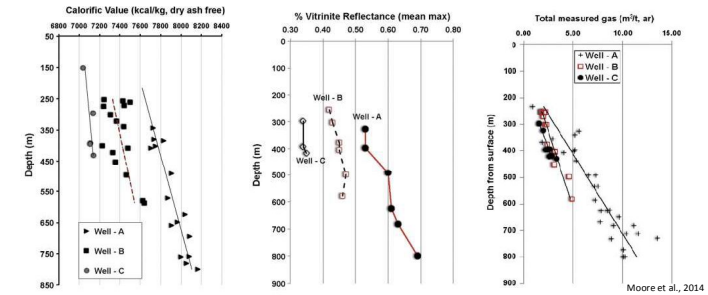
Moore, 2010

Coal properties & Gas content



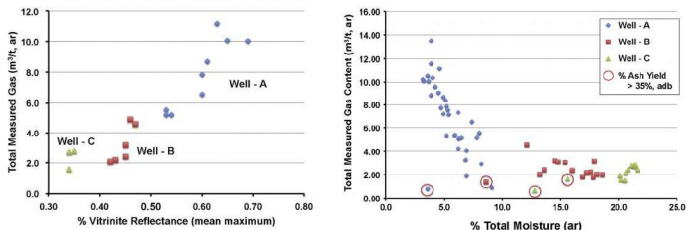
Moore et al., 2014

Coal properties & Gas content



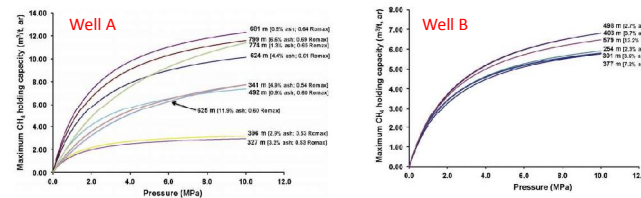
Moore et al., 2014

Coal properties & Gas content



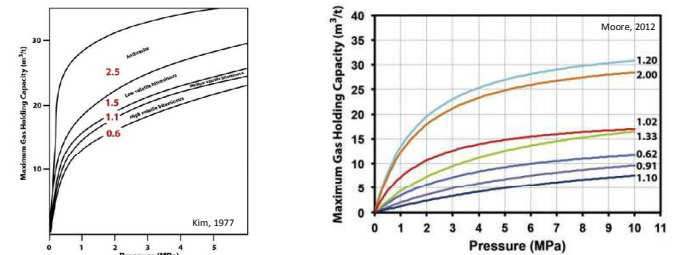
Moore et al., 2014

Adsorption isotherms



Moore et al., 2014

Adsorption isotherms related to rank



Rank-gas relationships shown are in general correct, they are far from universally observed

Gas in-place (GIP)

$$\text{GIP} = A \times \text{CT} \times d \times \text{GC}$$

- A : area or distribution of coal being estimated
- CT: coal thickness
- d : density
- GC: gas content

Moore, 2010

End