



Universitas Gadjah Mada	Universitas Gadjah Mada	Universitas Gadjah Mada
Hydrophobic - Hydrophilic	Wettability	Wettability
Hydrophilic Surface         Hydrophilic Surfa	<ul> <li>the preference of a solid to be in contact with one fluid rather than another (Abdullah et al., 2007)</li> <li>the ability of a fluid phase to preferentially wet a solid surface in the presence of a second immiscible phase</li> <li>wettability is generally classified into three categories: <ol> <li>The reservoir is said to be water wet; that is, water preferentially wets the reservoir rock, when the contact angle (θ)between the rock and water is less than 90°,</li> <li>neutral wettability case would exist at a contact angle of 90°, and</li> <li>oil wet occurs at a contact angle greater than 90°.</li> </ol> </li> </ul>	Rock       Rock       Rock       Image: Source of the so
Ferian - TGL2015 Physical Chemistry Multiphase phenomena- 7	Ferian -TGL2015 Physical Chemistry Multiphase phenomena- 8	Ferian -TGL2015 Physical Chemistry Multiphase phenomena- 9

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$\sigma_{OS} - \sigma_{WS} = \sigma_{OW} \cos \theta$ where: $\sigma_{w}$ = interfacial energy between oil and solid, dyne/cm; $\sigma_{w}$ = interfacial energy or interfacial energy of the water, dyne/cm; $\sigma_{w}$ = interfacial energy between oil and solid, dyne/cm; $\sigma_{w}$ = interfacial energy between oil and solid, dyne/cm; $\sigma_{w}$ = interfacial energy between oil and solid, dyne/cm; $\sigma_{w}$ = interfacial energy or interfacial tension, between oil and water, dyne/cm; $\theta$ = contact angle at oil-water-solid interface measured through the water phase, dref	Measurement of contact angle  Sessile drop method  Amott wettability test  United States Bureau of Mines (USBM) wettability test  Combined USBM-Amott Wettability test	<ul> <li>Proversities (sadgah Mada)</li> <li>Wettability curve</li> <li>The drainage is defined as a decrease in water saturation and imbibition as an increase.</li> <li>The forced drainage mechanism is characterized by larger pores tending to empty before smaller pores.</li> <li>The spontaneous imbibition mechanism is characterized by smaller pores filling before larger pores.</li> <li>The spontaneous imbibition is followed by forced displacement, a further increase in water saturation of oil, which by definition is a spontaneous imbibition of oil, which by definition is a spontaneous drainage process.</li> </ul>
Perior-TG22015 Physical Chemistry Multiphrae phenomena 10	Ferlan -TGL2015 Physical Chemistry Multiphase phenomeno-11	Morrow, 1990           Ferlan -TGL2015         Physical Chemistry         Multiphase phenomeno-12













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Coal flotation	Coal surface wettability	Contact angle vs. coal rank
• a technology by which coal is separated from mineral using the difference in the quality of adhering to air bubble in water	<ul> <li>Relationship between wettability and floatability is not straightforward,</li> <li>wettability → contact angle larger than zero (θ &gt; 0) can float</li> <li>it is generally assumed that wettability as expressed by contact angle is the most fundamental of floation-related surface property.</li> <li>low-rank coals which are not very hydrophobic float poorly,</li> <li>bituminous coals, if not oxidized, are hydrophobic and are easy to float</li> <li>coal matrix is assumed to be hydrophobic (to a varying degree), and this hydrophobicity is further modified by the presence of hydrophilic functional groups on the coal surface</li> <li>mineral matter impurities which are also hydrophilic.</li> </ul>	Water- Sensite Drops 10 10 10 10 10 10 10 10 10 10
Ferian - TGL2015 Physical Chemistry Multiphase phenomeno-31	Ferian - TGL2015 Physical Chemistry Multiphase phenomena-32	Ferian -TGL2015 Physical Chemistry Multiphose phenomeno-33

